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2. How many moles of Fe^{2+} ion can be oxidized by 1.0×10^{-3} moles MnO_4^- ion in the reaction in Question 1?
- 3.

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+ 4H₂O

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Advance Study Assignment: Rate of Reactions
II: A Clock Reaction? 1. A student studied
the clock reaction described in this
experiment. she set up Reaction Mixture 3 by
mixing 10 mL 0.010 M KI, 10 mL 0.0010
Na₂S₂O₃, 20 mL 0.040 M KBrO₃, and 10 mL M
HCl. Advance Study Assignment: Rate of
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Assignment Experiment 20 Experiment 20
Advance Study Assignment: The iodination of
Acetone 1. In a reaction involving the
iodination of acetone, the following volumes
were used to make up the reaction mixture:
5 mL. 4.0 M acetone+10 mL. 1.0 M HCl+10
mL. 0.0050 M I₂ +25 mL ...

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