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continued _____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. _____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH

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of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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